

Unofficial Solutions of “Solar Energy” Tutorial 6 (WT2021/22)

This is a rewrite of the sample solution (that is not handed out to the students) taken from a hybrid tutorial.

Exercise 1

- a. False
- b. True
- c. False
- d. False

Exercise 2

The Lambert-Beer law describes how the intensity of the light beam decays as it propagates through an absorbing medium with a certain absorption coefficient $\alpha(\lambda)$:

$$I(d) = I_0 \cdot \exp(-\alpha(\lambda) \cdot d)$$

First the absorption coefficient is taken from the figure, after converting the energy of the photons into wavelength via:

$$\lambda(\text{nm}) = \frac{h \cdot c}{q \cdot E_{\text{ph}}} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{1,602 \cdot 10^{-19} \text{ C} \cdot 1,55 \text{ eV}} = 800 \text{ nm}$$

Taking into account that the y-axis is plotted in a logarithmic scale, the absorption coefficient can be estimated and the required thickness of each material can be calculated:

$$\text{GaAs: } \alpha(800 \text{ nm}) = 1,5 \cdot 10^4 \frac{1}{\text{cm}}, \quad d = -\frac{\ln(0,1)}{1,5 \cdot 10^4 \frac{1}{\text{cm}}} = 1,5 \mu\text{m}$$

$$\text{InP: } \alpha(800 \text{ nm}) = 4 \cdot 10^4 \frac{1}{\text{cm}}, \quad d = -\frac{\ln(0,1)}{4 \cdot 10^4 \frac{1}{\text{cm}}} = 0,58 \mu\text{m}$$

$$\text{GaAs: } \alpha(800 \text{ nm}) = 6 \cdot 10^4 \frac{1}{\text{cm}}, \quad d = -\frac{\ln(0,1)}{6 \cdot 10^4 \frac{1}{\text{cm}}} = 0,38 \mu\text{m}$$

$$\text{GaAs: } \alpha(800 \text{ nm}) = 9 \cdot 10^2 \frac{1}{\text{cm}}, \quad d = -\frac{\ln(0,1)}{9 \cdot 10^2 \frac{1}{\text{cm}}} = 25,6 \mu\text{m}$$

Exercise 3

- a. Answer: $E_G = 1,9 \text{ eV}$

To calculate the bandgaps, we should take into account the wavelength until which a certain junction absorbs light.

Junction A absorbs light (has a non-zero EQE) until $\lambda = 650 \text{ nm}$. The wavelength can be converted into energy by using the relation:

$$E_G = \frac{h \cdot c}{\lambda \cdot q} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{650 \cdot 10^{-9} \text{ m} \cdot 1,602 \cdot 10^{-19} \text{ C}} = 1,9 \text{ eV}$$

b. Answer: $E_G = 1,46\text{eV}$

Junction B absorbs light (has a non-zero EQE) until $\lambda = 850\text{nm}$. The wavelength can be converted into energy by using the relation:

$$E_G = \frac{h \cdot c}{\lambda \cdot q} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{850 \cdot 10^{-9} \text{ m} \cdot 1,602 \cdot 10^{-19} \text{ C}} = 1,46\text{eV}$$

c. Answer: $E_G = 0,99\text{eV}$

Junction C absorbs light (has a non-zero EQE) until $\lambda = 1250\text{nm}$. The wavelength can be converted into energy by using the relation:

$$E_G = \frac{h \cdot c}{\lambda \cdot q} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{1250 \cdot 10^{-9} \text{ m} \cdot 1,602 \cdot 10^{-19} \text{ C}} = 0,99\text{eV}$$

d. Answer: C

In a multi-junction solar cell, the cell with the highest band gap is placed at the top, and the cell with the lowest band gap is placed at the bottom. Therefore, junction A acts as the top cell, junction B as the middle cell, and junction C as the bottom cell.

e. Answer: $J_{SC_A} = 10,42 \frac{\text{mA}}{\text{cm}^2}$, $J_{SC_B} = 12,10 \frac{\text{mA}}{\text{cm}^2}$, $J_{SC_C} = 17,92 \frac{\text{mA}}{\text{cm}^2}$

The short circuit current density can be calculated as:

$$J_{SC} = q \cdot \int \text{EQE}(\lambda) \cdot \Phi_{ph,\lambda} d\lambda = q \cdot \text{EQE} \cdot \Phi_{ph}$$

Junction A has an EQE of 0,7 between a wavelength of 300nm and 650nm. This gives:

$$J_{SC_A} = 1,602 \cdot 10^{-19} \text{ C} \cdot 0,7 \cdot 9,3 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} = 104,2 \frac{\text{A}}{\text{m}^2} = 10,42 \frac{\text{mA}}{\text{cm}^2}$$

Junction B has an EQE of 0,9 between a wavelength of 650nm and 850nm. This gives:

$$J_{SC_B} = 1,602 \cdot 10^{-19} \text{ C} \cdot 0,9 \cdot 8,4 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} = 121,0 \frac{\text{A}}{\text{m}^2} = 12,10 \frac{\text{mA}}{\text{cm}^2}$$

Junction C has an EQE of 0,8 between a wavelength of 850nm and 1250nm. This gives:

$$J_{SC_C} = 1,602 \cdot 10^{-19} \text{ C} \cdot 0,8 \cdot 1,4 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} = 179,2 \frac{\text{A}}{\text{m}^2} = 17,92 \frac{\text{mA}}{\text{cm}^2}$$

f. Answer: $\eta = 17\%$

The efficiency is calculated as:

$$\eta = \frac{J_{SC} \cdot V_{OC} \cdot FF}{P_{in}}$$

The open circuit voltages are:

$$V_{OC_A} = \frac{1,9\text{eV}}{2q} = 0,95\text{V}, \quad V_{OC_B} = \frac{1,46\text{eV}}{2q} = 0,73\text{V}, \quad V_{OC_C} = \frac{0,99\text{eV}}{2q} = 0,495\text{V}$$

Since the cells are connected in series, the short circuit current density of the multi-junction cell will be limited by the lowest short circuit density. On the other hand, the open circuit voltages will add up. Thus:

$$\begin{aligned} \eta &= \frac{J_{SC} \cdot V_{OC} \cdot FF}{P_{in}} = \frac{J_{SC_A} \cdot (V_{OC_A} + V_{OC_B} + V_{OC_C}) \cdot FF}{P_{in}} \\ &= \frac{10,42 \frac{\text{mA}}{\text{cm}^2} \cdot (0,95\text{V} + 0,73\text{V} + 0,495\text{V}) \cdot 0,75}{100 \frac{\text{mW}}{\text{cm}^2}} = 0,1966 \triangleq 17\% \end{aligned}$$

Exercise 4

a. Answer: spectral range A

The bandgap is a measure of the lowest photon energy that can be absorbed in the material. The corresponding wavelength is calculated as:

$$\lambda = \frac{h \cdot c}{q \cdot E_G} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{1,602 \cdot 10^{-19} \text{ C} \cdot 2,0 \text{ eV}} = 620 \text{ nm}$$

This corresponds to spectral range A.

b. Answer: $J_{SC} = 10,4 \frac{\text{mA}}{\text{cm}^2}$

The short circuit current density can be calculated as:

$$J_{SC} = q \cdot \int \text{EQE}(\lambda) \cdot \Phi_{ph\lambda} d\lambda = q \cdot \text{EQE} \cdot \Phi_{ph}$$

The solar cell has an EQE of 0,65 between a wavelength of 0 nm and 620 nm. This gives:

$$J_{SC} = 1,602 \cdot 10^{-19} \text{ C} \cdot 0,65 \cdot 10 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} = 104 \frac{\text{A}}{\text{m}^2} = 10,4 \frac{\text{mA}}{\text{cm}^2}$$

c. Answer: $\eta = 8,3\%$

The efficiency is calculated as:

$$\eta = \frac{J_{SC} \cdot V_{OC} \cdot FF}{P_{in}}$$

The open circuit voltage is calculated with the given equation:

$$V_{OC_A} = \frac{2 \text{ eV}}{2q} = 1 \text{ V}, \quad V_{OC_B} = \frac{1,46 \text{ eV}}{2q} = 0,73 \text{ V}, \quad V_{OC_C} = \frac{0,99 \text{ eV}}{2q} = 0,495 \text{ V}$$

Therefore, the efficiency is:

$$\eta = \frac{J_{SC} \cdot V_{OC} \cdot FF}{P_{in}} = \frac{10,4 \frac{\text{mA}}{\text{cm}^2} \cdot 1 \text{ V} \cdot 0,8}{100 \frac{\text{mW}}{\text{cm}^2}} = 0,083 \triangleq 8,3\%$$

d. Answer: spectral range B

In order for the photon emitted by the upconverter to be absorbed in the a-SiC:H layer, the sum of the energy of the two (lower-energy) photons must be at least equal to the a-SiC:H bandgap. Therefore, the minimum energy that each of the two photons must carry is:

$$\frac{2 \text{ eV}}{2 \text{ photons}} = 1 \frac{\text{eV}}{\text{ph}}$$

$$\lambda = \frac{h \cdot c}{q \cdot E_G} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{1,602 \cdot 10^{-19} \text{ C} \cdot 1 \text{ eV}} = 1240 \text{ nm}$$

Since we assume that photons up to 620 nm are absorbed in the a-SiC:H layer, the photons can be up-converted in spectral range B.

- e. Answer: $J_{SC} = 16 \frac{\text{mA}}{\text{cm}^2}$, $\eta = 10,4\%$

The short circuit current density can be calculated as:

$$J_{SC} = q \cdot \int \text{EQE}(\lambda) \cdot \Phi_{\text{ph}_\lambda} d\lambda = q \cdot \text{EQE} \cdot \left(1 \cdot \Phi_{\text{ph}_A} + \frac{1}{2} \cdot \Phi_{\text{ph}_B} \right)$$

Take into account that the factor $\frac{1}{2}$ in front of the second integral is due to the fact that the upconverter converts two low-energy photons into one high-energy photon (we lose half of the photon flux).

The solar cell has an EQE of 0,65 between a wavelength of 0 nm and 1240 nm. This gives:

$$J_{SC} = 1,602 \cdot 10^{-19} \text{ C} \cdot 0,65 \cdot \left(10 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} + \frac{1}{2} \cdot 11 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} \right) = 160 \frac{\text{A}}{\text{m}^2} = 16 \frac{\text{mA}}{\text{cm}^2}$$

The efficiency is calculated as:

$$\eta = \frac{J_{SC} \cdot V_{OC} \cdot FF}{P_{in}} = \frac{16 \frac{\text{mA}}{\text{cm}^2} \cdot 1 \text{ V} \cdot 0,65}{100 \frac{\text{mW}}{\text{cm}^2}} = 0,104 \triangleq 10,4\%$$

- f. Answer: spectral range C

In order for the photon emitted by the upconverter to be absorbed in the a-SiC:H layer, the sum of the energy of the two (lower-energy) photons must be at least equal to the a-SiC:H bandgap. Therefore, the minimum energy that each of the two photons must carry is:

$$\frac{2 \text{ eV}}{3 \text{ photons}} = 0,667 \frac{\text{eV}}{\text{ph}}$$

$$\lambda = \frac{h \cdot c}{q \cdot E_G} = \frac{6,626 \cdot 10^{-34} \text{ Js} \cdot 2,998 \cdot 10^8 \frac{\text{m}}{\text{s}}}{1,602 \cdot 10^{-19} \text{ C} \cdot 0,667 \text{ eV}} = 1860 \text{ nm}$$

Since we assume that photons up to 620 nm are absorbed in the a-SiC:H layer and that photons up to 1240 nm are converted in upconverter 1, the photons can be up-converted in spectral range C.

- g. Answer: $J_{SC} = 18 \frac{\text{mA}}{\text{cm}^2}$, $\eta = 11,7\%$

The short circuit current density can be calculated as:

$$J_{SC} = q \cdot \int \text{EQE}(\lambda) \cdot \Phi_{\text{ph}_\lambda} d\lambda = q \cdot \text{EQE} \cdot \left(1 \cdot \Phi_{\text{ph}_A} + \frac{1}{2} \cdot \Phi_{\text{ph}_B} + \frac{1}{3} \cdot \Phi_{\text{ph}_C} \right)$$

Take into account that the factor $\frac{1}{3}$ in front of the third integral is due to the fact that the upconverter converts three low-energy photons into one high-energy photon (we lose two-third of the photon flux).

The solar cell has an EQE of 0,65 between a wavelength of 0 nm and 1860 nm. This gives:

$$J_{SC} = 1,602 \cdot 10^{-19} \text{ C} \cdot 0,65 \cdot \left(10 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} + \frac{1}{2} \cdot 11 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} + \frac{1}{3} \cdot 6 \cdot 10^{20} \frac{1}{\text{m}^2 \cdot \text{s}} \right) = 180 \frac{\text{A}}{\text{m}^2} = 18 \frac{\text{mA}}{\text{cm}^2}$$

The efficiency is calculated as:

$$\eta = \frac{J_{SC} \cdot V_{OC} \cdot FF}{P_{in}} = \frac{18 \frac{\text{mA}}{\text{cm}^2} \cdot 1 \text{ V} \cdot 0,65}{100 \frac{\text{mW}}{\text{cm}^2}} = 0,117 \triangleq 11,7\%$$